### NADH oxidase

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## **Summary**

NADH oxidase catalyzes the oxidation of NADH to NAD with water as a co-product. This protocol describes a direct enzyme assay measuring the quantity of NADH consumed. The co-substrate is oxygen, and therefore the assay mixture must contain dissolved oxygen.

## Solutions Required

- 1. 125 mM phosphate buffer pH = 7.0 0.827 g KH<sub>2</sub>PO<sub>4</sub> plus 2.349 g K<sub>2</sub>HPO<sub>4</sub>·3H2O in 100 mL DI water
- 2. 2.9 mM NADH must be prepared fresh per assay need 0.0002 g / 0.05 mL
- 3. 5 mM dipotassium EDTA (FW = 404.47) 0.0041 g/ 2 mL

# Preparation of Cell Extract

Follow general protocol **Preparation of Cell Extract**.

- 1. Centrifuge sufficient cells so that the volume diluted down to 5 mL would give an optical density of 20-30. For example, for a broth of OD=1, use 100 mL. For a broth of OD=10, use 10 mL.
- 2. After first pelletization of cells, resuspend in 5-15 mL of tricine buffer.
- 3. After second pelletization of cells, resuspend in 5 mL of tricine buffer, and break with French Press.

## Spectrophotometer

Turn on the ultraviolet bulb on the spectrophotometer (Beckman DU50) and wait 30 minutes for warm-up. Select the kinetics-time window on the instrument. Load the method "A:/nadh". This method has a run-time of 60 s, a temperature of 37°C, a wavelength of 340 nm and uses 2 autosamplers.

#### Procedure

1. For each assay, prepare a cocktail shown in the following table into one UV-translucent cuvettes, and keep on ice. There is no control. "Blank" the spectrophotometer with DI water.

Solution	Volume (µL) added to:
DI H <sub>2</sub> O	400
Phosphate	400
EDTA	100
NADH	50

- 2. Directly from the ice when ready to commence the assay, place the cuvette (containing 950 µL†) into the spectrophotometer holder.
- 3. Wait 10 minutes to allow the temperature of the solutions in the cuvettes to equilibrate.
- 4. Aspirate air into the cuvette with a pipettor.
- 5. Simultaneously add 50 µL† of the cell extract to the cuvettes.
- 6. To mix solutions, immediately and simultaneously aspirate and dispense the contents of the cuvettes with a pipettor. Mix the solutions in this way ten times. (Count!)
- 7. Promptly depress "start" on the monitor.
- 8. Record the rates for the two (control and experimental) cuvettes.

† Dilution of the cell extract may be adjusted so that change in absorbance is between about 0.05 and 0.7 AU in one minute. This dilution should be accomplished externally in a microcentrifuge tube (for example, by adding 50  $\mu$ L of cell extract to 950  $\mu$ L DI water to achieve a dilution of 20). The volume of 50  $\mu$ L should always be used in the enzyme assay mixture.

# Calculation of Activity

One unit (U) of NADH oxidase is defined as the amount of enzyme required to oxidase 1.0 µmole of NADH in one minute.

1. 
$$dA/dt (min^{-1}) = [Rate]_{experimental} - [Rate]_{control} = dA/dt$$

2. Activity = 
$$\frac{1000 \times TV \times D \times dA/dt}{\varepsilon \times V \times CF}$$

Activity: Volumetric Activity (U/L)

TV: Total volume in cuvette (1000 µL)

D: Dilution of the cell extract. (For example, if 50 µL of cell extract were add to 950 µL

DI water prior to using a volume of cell extract in the assay, then D=20)

V: Volume of cell extract used (50 µL)

ε: Molar extinction coefficient for NADH (6.22 L/mmol for a path length of 1.0 cm)

CF: Concentration Factor of cell extract (For example, if a 100 mL sample is

concentrated to a 2 mL volume for the French Press, then CF=50)

3. Specific Activity = 
$$\frac{Activity}{Protein Concentration}$$
 1

Activity: Volumetric Activity, as calculated in #2 above (U/L)

Protein Concentration: Protein concentration, as calculated in protocol Total Protein

**Concentration** (mg/L)

Specific Activity: (U/mg protein)