PEP Carboxykinase

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Summary

PEP carboxykinase catalyzes the carboxylation of PEP to oxaloacetate with the generation of ATP. In this coupled assay, the ATP generated by this reaction acts as a substrate in the conversion of 3-phosphoglycerate to 1,3-diphosphoglycerate via the enzyme 3-phosphoglycerate phosphokinase. The 1,3-diphosphoglycerate so generated is then converted to glyceraldehyde 3-phosphate via the enzyme glyceraldehyde 3-phosphate dehydrogenase and NADH, which is supplied in the reaction mixture. The disappearance of NADH is what is actually measured.

Solutions Required

- 1. 500 mM tris-HCl pH = 6.6
- 2. 350 mM NaHCO₃

can be prepared in stock solution and stored at room temperature.

3. 160 mM MgCl₂

can be prepared in stock solution and stored at room temperature.

4. 6 mM NADH

can be prepared in stock solution and stored at -20°C

5. 20 mM dithiothreitol (DTT)

can be prepared in stock solution and stored at -20°C

6. 0.2 M ADP

can be prepared in stock solution and stored at -20°C

7. 36 mM 3-phosphoglycerate

must be prepared fresh

8. 50 mM PEP

must be prepared fresh

9. 3-Phosphoglycerate phosphokinase (PGPK)

suspension as purchased from Sigma (P7634)

prepare just prior to use a solution which contains $4 \text{ U}/100 \mu\text{L}$. A $100 \text{ }\mu\text{L}$ will be needed for each assay (50 μL for experimental and 50 μL for control).

10. Glyceraldehyde 3-phosphate dehydrogenase (GPDH)

suspension as purchased from Sigma (G9263, 92 U/mg)

prepare just prior to use a solution which contains 4 U/100 μ L. A 100 μ L will be needed for each assay (50 μ L for experimental and 50 μ L for control).

Preparation of Cell Extract

Follow general protocol **Preparation of Cell Extract**.

- 1. Centrifuge sufficient cells so that the volume diluted down to 2 mL would give an optical density of 40-50. For example, for a broth of OD=1, use 100 mL. For a broth of OD=10, use 10 mL.
- 2. After first pelletization of cells, resuspend in 5-15 mL of 4°C 100 mM tris-HCl buffer.
- 3. After second pelletization of cells, resuspend in 2 mL of 4°C 100 mM tris-HCl buffer, and break with French Press.

Spectrophotometer

Turn on the ultraviolet bulb on the spectrophotometer (Beckman DU50) and wait 10 minutes for warm-up. Select the kinetics-time window on the instrument. Load the method "A:/ppc". This method has a run-time of 60 s, a temperature of 37°C, a wavelength of 340 nm and uses 2 autosamplers.

Procedure

1. For each assay, prepare the two cocktails shown in the following table into two separate UV-translucent cuvettes, and keep them on ice.

	Volume (µL) added to:	
Solution	Control	Experimental
DI H ₂ O	200	100
tris	200	200
NaHCO ₃	100	100
$MgCl_2$	100	100
DTT	50	50
ADP	50	50
3-phosphoglycerate	50	50
PEP	0	100
NADH	50	50
PGPK	50	50
GPDH	50	50

- 2. Directly from the ice when ready to commence the assay, place the two cuvettes (each containing 900 µL†) into the spectrophotometer holder (position #1 for control, position #2 for experimental).
- 3. Wait 10 minutes to allow the temperature of the solutions in the cuvettes to equilibrate.
- 4. Blank the spectrophotometer using water ("Read samples").
- 5. Simultaneously add 100 µL† of the cell extract to the cuvettes.

- 6. To mix solutions, immediately and simultaneously aspirate and dispense the contents of the cuvettes with a pipettor. Mix the solutions in this way ten times. (Count!)
- 7. Promptly depress "start" on the monitor.
- 8. Record the rates for the two (control and experimental) cuvettes.
- † Our experience is that the cell extract has to be diluted by a factor of about 20 when using overexpressed PCK. This dilution should be accomplished in a microcentrifuge tube (for example, by adding 50 μ L of cell extract to 950 μ L DI water). The volume of 100 μ L should always be used in the enzyme assay mixture.

Calculation of Activity

One unit (U) of PEP carboxykinase activity is defined as the amount of enzyme required to produce $1.0~\mu mole$ of oxaloacetate in one minute. This quantity is proportional to the amount required to generate $1.0~\mu mole$ ATP, and to the amount required to generate $1.0~\mu mole$ NADH via this coupled enzyme assay.

1.
$$dA/dt (min^{-1}) = [Rate]_{experimental} - [Rate]_{control} = dA/dt$$

2. Activity =
$$\frac{1000 \times TV \times D \times dA/dt}{\varepsilon \times V \times CF}$$

Activity: Volumetric Activity (U/L)

TV: Total volume in cuvette (1000 µL)

D: Dilution of the cell extract. (For example, if 50 µL of cell extract were add to 950 µL

DI water prior to using a volume of cell extract in the assay, then D=20)

V: Volume of cell extract used (100 µL)

ε: Molar extinction coefficient for NADH (6.22 L/mmol for a path length of 1.0 cm)
CF: Concentration Factor of cell extract (For example, if a 100 mL sample is

concentrated to a 2 mL volume for the French Press, then CF=50)

3. Specific Activity =
$$\frac{Activity}{Protein Concentration}$$

Activity: Volumetric Activity, as calculated in #2 above (U/L)

Protein Concentration: Protein concentration, as calculated in protocol **Total Protein**

Concentration (mg/L)

Specific Activity: (U/mg protein)

Reference

P. Kin, M. Laivenieks, C. Vieille, J. G. Zeikus (2004) Effect of overexpression of *Actinobacillus succinogenes* phosphopyruvate carboxykinase on succinate production in *Escherichia coli*, Appl.

Environ. Microbiol. 70(2):1238-1241.